Evidence for a Plastic Crystalline Phase in Closo-N-Methylphosphorimide and Closo-N-Methylarsenimide

M. POSTEL, F. CASABIANCA and J. G. RIESS

Laboratoire de Chimie Minérale Moléculaire, Equipe de Recherche Associée au CNRS, Institut de Mathématiques et Sciences Physiques, Parc Valrose, 06034 Nice Cedex, France

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One striking feature in phosphorus chemistry is its proclivity to form *closo*-type structures (or so-called "cage compounds"*), as for example in its oxides, sulphides, oxysulphides, imides, thioimides, and so on.¹ Several among these have T_d molecular symmetry and globular, nearly spherical shapes. Therefore the possibility exists that these molecules should easily recover their degrees of freedom in the solid and exhibit plastic phases² in which the molecules would undergo rapid restricted rotation in their unit cell, as well as self-diffusion in the crystalline solid. Such behaviour is well documented in the case of the structurally related adamantane and analogous compounds.³

We now wish to report the observation of plastic phases for $P_4(NMe)_6$ and $As_4(NMe)_6$.

A proton nmr signal becomes detectable for solid $P_4(NMe)_6^{1c,4}$ on a High Resolution 60MHz spectrometer (JEOL C-60 HL) from 40 °C upwards, which already supposes considerable rapid reorientational movement in the solid at that temperature. The peak width at half height abruptly changes from *ca.* 4,700 Hz (1.12 gauss) to *ca.* 400 Hz (0.1 gauss) within one degree at 56 °C when the sample is heated. This reflects a drastic change in molecular freedom, which we assign to a plastic phase transition T_P , above which the molecules suddenly begin to rotate, while maintaining their positional order in the solid.

This interpretation is further supported by the following facts: (i) there is a change in aspect of the crystalline material, which suddenly becomes translucent; (ii) differential scanning calorimetry (DSC) shows a sharp first order-transition at 330 ± 0.5 °K upon heating ($\Delta H = 7.57$ KJ mol⁻¹, $\Delta S = 23.01$ J mol⁻¹ °K⁻¹). Upon cooling this transition is found at 323.5 °K, which denotes hysteresis, a common feature of plastic phase transitions; (iii) variable-

temperature X-ray powder-diffraction shows a change in pattern at 323 ± 5 °K from monoclinic⁵ to cubic upon heating. The diffraction lines, which are sharp and numerous for the "rigid" crystal, become few (5 reflexions) and less well-defined for the plastic crystal.⁶ The original pattern is slowly recovered upon cooling, with lines corresponding to the cubic system still distinguishable down to 308 °K, again showing that, although it is reversible, there is an appreciable hysteresis in the process; (iv) the low entropy of melting ($\Delta H = 1.9 \text{ KJ mol}^{-1}$, $\Delta S = 4.8 \text{ J}$ mol⁻¹ °K⁻¹ by DSC at T_m = 395 °K) is also characteristic of a plastic crystalline phase;² (v) the ³¹P nmr signal, which becomes detectable at *ca.* 50 °C upwards, shows the same abrupt decrease in line width at 56° ± 1 °C.

The existence of a plastic phase transition for $P_4(NMe)_6$ should be considered collaterally with the relatively low symmetry of the space group (monoclinic, $P2_1/c$, two molecules per unit cell)⁵ in which the highly symmetric molecule crystallises, and above all, with the disorder which has been found in the low temperature crystalline phase of related *closo*-compounds, which suggests that the globular shape of the molecule presents too few asperities to induce highly symmetric order, even in the rigid phase.

A second, less pronounced and smoother transition was also observed in the ¹H nmr line-width experiment: between *ca.* 338 and 348 °K, the signal is observed to narrow from 350 to 120 Hz, a change which we assign to self-diffusion,⁸ which then becomes rapid enough to influence the nmr line shapes. On further heating the spectral resolution slowly improves until the fine structure (the two symmetric lines corresponding to N = $2(J_{AX} + J_{AX'}) =$ 32 Hz of the second-order spectrum^{1c,4}) becomes apparent. There is no change in the nmr at the melting point.

A third transition, at lower temperature, is detected by DSC (309 °K, first-order upon cooling, $\Delta H = 3.01$ KJ mol⁻¹, $\Delta S = 9.7$ J mol⁻¹ K⁻¹; higher order upon heating) which we tentatively assign to the rotation of the molecule about one of its C₃ axes. The fact that this process is higher order upon heating means that the onset of the rotation is concerted.¹⁰

Similar behaviour was found for solid As₄(NMe)₆, which exhibits a large plastic range ($T_p = 346^\circ$, $T_m = 393^\circ$ K) as shown by an abrupt narrowing of the ¹H nmr line width at 346 °K.

^{*}A rather misleading denomination, since there is no room whatever in the "cage".)

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